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CLAIMS:

- 1. A solution of a precursor for the pyrolytic formation of a tin-containing coating comprising stannic hydroxychloride.
- 5 2. The solution of claim 1 wherein said stannic hydroxychloride is a non-stoichiometric compound expressed by the formula Sn(OH)_{2+x}Cl_{2-x}.nH₂O, wherein the Cl:Sn molar ratio is comprised between 1 and 1.9.
 - 3. The solution of claim 1 wherein said stannic hydroxychloride is a compound expressed by the formula $SnO(H_2O)_nR_{2-x}Cl_x$, being R a preferably organic substituent.
 - 4. The solution of claim 3 wherein the CI:Sn molar ratio is comprised between 1 and 1.9.
 - 5. The solution of claim 3 or 4 wherein R is the acetic group (CH₃COO-)
 - 6. The solution of any one of the previous claims further comprising a precursor of at least one noble metal.
 - 7. The solution of claim 6 wherein said precursor of at least one noble metal is a chlorinated precursor of iridium or ruthenium.
 - 8. The solution of claim 7 wherein said chlorinated precursor of iridium is H₂IrCl₆.
- 9. An anode provided with an electrocatalytic coating comprising tin, preferably tetravalent and in form of mixed oxide, prepared by pyrolysis of a solution of any one of the previous claims.
 - 10. The anode of claim 9 wherein the electrocatalytic coating is deposited on a substrate of a valve metal, preferably titanium or a titanium alloy.
- 25 11. The anode of claim 10 wherein a ceramic pre-layer is interposed between the coating and said substrate.
 - 12. The anode of claim 11 wherein said ceramic pre-layer comprises titanium dioxide.
- 13. The anode of any one of claims 9 to 12 wherein said coating has electrocatalytic properties toward the chlorine evolution reaction and said at least one noble metal is ruthenium.

- 14. The anode of any one of claims 9 to 12 wherein said coating has electrocatalytic properties toward the oxygen evolution reaction and said at least one noble metal is iridium.
- 15. A method for the manufacturing of a precursor solution for the pyrolytic formation of a tin-containing coating comprising the addition of hydrogen peroxide to a stannous chloride solution, optionally under temperature and redox potential control.
 - 16. The method of claim 15 wherein the Cl:Sn ratio in the solution is decreased by subsequent reduction of metallic tin and further addition of hydrogen peroxide, optionally under temperature and redox potential control.
 - 17. The method of claim 15 or 16 wherein said stannous chloride solution further contains a precursor of an organic substituent.
 - 18. The method of claim 17 wherein said precursor of an organic substituent is acetic acid.
- 19. A method for the manufacturing of an electrode comprising the application of the solution of anyone of claims 1 to 8 to a substrate of a valve metal, preferably titanium or titanium alloy optionally provided with a ceramic pre-layer followed by the execution of a thermal treatment.
 - 20. The method of claim 19 wherein said application of the solution is effected in multiple coats, each followed by a thermal treatment.
 - 21. The method of claim 19 or 20 wherein said thermal treatment is a pyrolysis at a temperature comprised between 350 and 800°C, optionally preceded by a drying at a temperature comprised between 80 and 200°C.

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